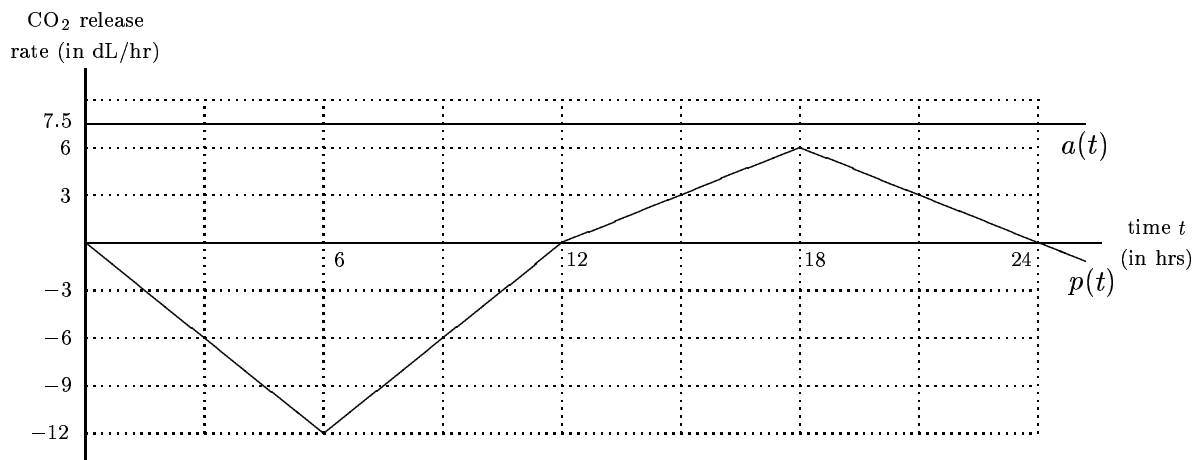


8. (14 points) A team of biologists seeking to develop alternate solutions to the use of pesticides proposes the following experiment. A plant infested by a large colony of aphids (small insects), is placed in a container originally saturated in dioxygen ( $O_2$ ). They hope that the parasites can be suffocated by the  $CO_2$  produced by their natural activities (e.g. breathing) and those of the plant (e.g. photosynthesis). Make the simplifying assumptions, which are approximations, that:

- (i) The container is sealed; no molecules enter or leave.
- (ii) The sum of the volumes of  $O_2$  and  $CO_2$  in the container is constant and equal to 117 deciliters (dL); i.e. 11.7 liters.
- (iii) At the start of the experiment ( $t = 0$ ), there is no  $CO_2$  in the container.

The light in the room is adjusted so as to mimick a “perfect” 24-hour period. The parasites produce  $CO_2$  at a constant rate of 7.5 dL per hour. When the light is turned on, the plant absorbs  $CO_2$  and releases  $O_2$ . When the light is turned off, it does the opposite; i.e. uses  $O_2$  and produces  $CO_2$ . The rates of release in the container of  $CO_2$  (in dL per hour) for the parasites,  $a(t)$ , and the plant,  $p(t)$ , are plotted below. Note that negative rates correspond to absorption.



(a) By approximately how much does the volume  $V(t)$  of  $CO_2$  in the container change in the small time interval between  $t$  and  $t + \Delta t$ ? Express your answer in terms of  $\Delta t$ , and of the rates  $p(t)$  and  $a(t)$ .

*The production of  $CO_2$  in the tank comes from both the aphids and the plant. So the total rate at which  $CO_2$  is released is given by  $a(t) + p(t)$ . Thus, in the small time interval from  $t$  to  $t + \Delta t$ , the (small) change in the volume of  $CO_2$  is :*

$$\Delta V = (a(t) + p(t)) \Delta t .$$

Solution continued on next page.

Solution continued from previous page.

(b) Write an integral that gives the volume  $V(12)$  of  $\text{CO}_2$  in the container after 12 hours. Explain why your integral gives the value of  $V(12)$ . You will probably want to use the answer from (a) in your explanation.

We are told that there is no  $\text{CO}_2$  in the tank at time  $t = 0$ , so  $V(0) = 0$ . To find the volume of  $\text{CO}_2$  present in the tank after 12 hours, i.e.  $V(12)$ , we can “slice” the interval from  $t = 0$  to  $t = 12$  into small time intervals of length  $\Delta t$ . According to part (a), during each of these small time intervals, the change in the volume of carbon dioxide is  $\Delta V = (a(t) + p(t))\Delta t$ . Summing all those little contributions gives a Riemann sum. Letting  $\Delta t$  go to zero, the Riemann sum gives the integral:

$$V(12) = \int_0^{12} (a(t) + p(t)) dt.$$

(c) What is the value of  $V(12)$ ? Explain how you obtained your answer.

The integral in (b) represents the sum of the area under the graph of  $a(t)$  and the area under the graph of  $p(t)$ , between  $t = 0$  and  $t = 12$ . To find its value, it suffices to count the “boxes” on the chart. The scale on the figure shows that each box represents a volume of 9 dL of  $\text{CO}_2$ . Counting the boxes gives:

$$V(12) = (10 - 8) \times 9 = 18 \text{ deciliters.}$$

(d) When the container becomes saturated in  $\text{CO}_2$ , i.e. no  $\text{O}_2$  remains, the parasites suffocate to death, and the experiment is stopped. Decide whether that will happen within the 24-hour period; and, if you think it will, estimate this time.

We are told there are 117 dL of  $\text{O}_2$  available at  $t = 0$ . All aphids will thus die if the the volume of  $\text{CO}_2$  in the container reaches the value 117. To decide whether that will happen before the end of the experiment, i.e. before  $t = 24$  hours, we evaluate  $V(24)$  by counting the boxes as was done in (c). Doing so gives:

$$V(24) = (20 - 8 + 4) \times 9 = 144 > 117 \text{ deciliters.}$$

Accordingly, the aphids **will** indeed suffocate to death before the experiment ends.

Again counting the boxes, we find that will happen after **21 hours**. Indeed, at  $t = 21$  the total volume of  $\text{CO}_2$  in the container is:

$$V(21) = (17.5 - 8 + 3.5) \times 9 = 117 \text{ deciliters.}$$