

6. [14 points] Consider a chemical reaction in which two chemicals X and Y combine to form a new compound Z . We write $X + Y \rightarrow Z$. Then the speed of the reaction (that is, the rate at which the compound Z appears) is proportional to product of the concentrations of the compounds X and Y . Because one molecule of each of X and Y are used for each molecule of Z that is created, this results in the differential equation

$$\frac{dz}{dt} = \alpha(x_0 - z)(y_0 - z),$$

where z is the concentration of Z , α is the rate constant for the reaction and x_0 and y_0 are the initial concentrations of X and Y . If we initially have none of compound Z , the initial condition is $z(0) = 0$.

- a. [7 points] Suppose that $0 < \alpha < 1$ and $0 < x_0 < y_0$. Without solving the equation, determine what you expect the long-term concentration of Z will be by doing a qualitative analysis of the given equation. (While you may confirm your conclusions by speaking to the chemistry, your answer should be grounded in the analysis of the differential equation.)

- b. [7 points] Now suppose that $0 < \alpha < 1$ and $x_0 = y_0 > 0$. How does your analysis of the equation from (a) change? Explain by doing a similar analysis.