3. [15 points] A chemical reaction with two reagents (chemicals) in amounts r_1 and r_2 that may be converted from one to the other may be modeled the system of first-order differential equations

$$r'_{1} = -3r_{1} + 9r_{2}$$

$$r'_{2} = k r_{1} - r_{2} + f(t),$$

where f(t) is the rate at which the second reagent is being added to the reaction and k is a constant.

a. [5 points] Write down the second-order linear equation which has r_1 as its solution.

b. [5 points] If $f(t) = \cos(\omega t)$ is the dashed curve in the figure below, for what values of k, if any, could the long-term behavior of r_1 be that shown by the solid curve? Explain your answer.



c. [5 points] If $f(t) = A_0$, a constant, for what values of k, if any, could the phase portrait for this system be similar to that shown in the figure below? Explain your answer.

